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Memory Workout – Scholarship 13+ Science



Questions in bold are for scholarship students only

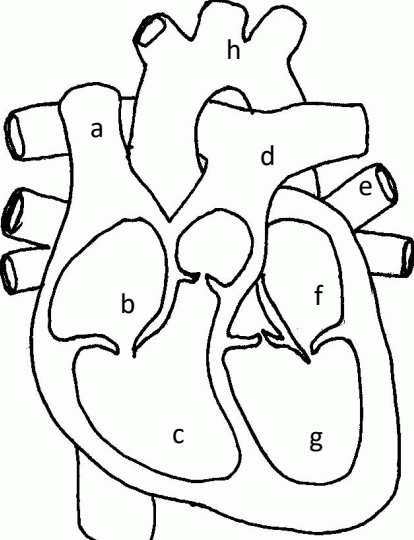
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| <p>Identify each labelled part of the diagram below:</p>  | <p>A – Vena Cava B – Right atrium C – Right ventricle D – Pulmonary artery E – Pulmonary vein F – Left atrium G – Left ventricle H – Aorta</p> | |
| <p>On which side of the heart does oxygenated blood flow?</p> | <p>The right</p> | |
| <p>On which side of the heart does deoxygenated blood flow?</p> | <p>The left</p> | |
| <p>What feature of the heart prevents blood from flowing in the wrong direction?</p> | <p>Valves</p> | |
| <p>What are the four components of blood?</p> | <ul style="list-style-type: none"> • Red blood cells • White blood cells • Platelets • Plasma | |
| <p>What is the function of the red blood cells?</p> | <p>To carry oxygen around the body</p> | |
| <p>What is the function of the white blood cells?</p> | <p>To fight diseases</p> | |
| <p>What is the function of the platelets?</p> | <p>To cause blood to clot, preventing bleeding</p> | |
| <p>What is the function of the plasma</p> | <p>It is the liquid part of blood which carries the other cells.</p> | |

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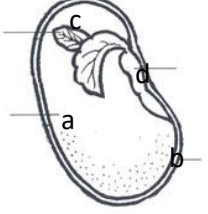
Biology – cellular respiration

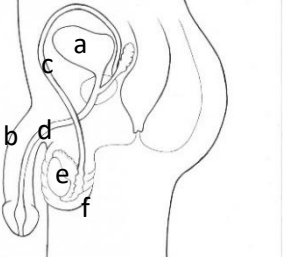
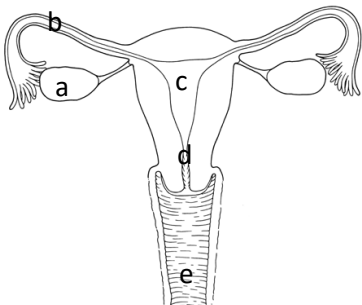
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| What is the word equation for photosynthesis? | Carbon dioxide + water → glucose + oxygen | | | | | | | | | | | | | | | | | | | |
| What is also required for photosynthesis to take place? | Light | | | | | | | | | | | | | | | | | | | |
| In which part of a plant does photosynthesis take place? | The leaves | | | | | | | | | | | | | | | | | | | |
| In which part of a plant cell does photosynthesis take place? | Chloroplast | | | | | | | | | | | | | | | | | | | |
| What is the name of the substance inside the chloroplast which allows photosynthesis to take place? | Chlorophyll | | | | | | | | | | | | | | | | | | | |
| Which colours of light are mostly absorbed by green leaves? | Red/orange | | | | | | | | | | | | | | | | | | | |
| What three things happens to the glucose after it has been made? | <ul style="list-style-type: none"> • It is converted to starch for storage • It is used in respiration • It is used for growth to become cell walls, seeds or fruits | | | | | | | | | | | | | | | | | | | |
| Which four factors may affect the rate of photosynthesis? | <ul style="list-style-type: none"> • Light intensity • Concentration of carbon dioxide • Temperature • Volume of water (although this is less important) | | | | | | | | | | | | | | | | | | | |
| What is the effect of increasing the light intensity on the rate of photosynthesis? | It will increase | | | | | | | | | | | | | | | | | | | |
| What is the effect of increasing the concentration of carbon dioxide on the rate of photosynthesis? | It will increase | | | | | | | | | | | | | | | | | | | |
| What is the effect of increasing the temperature on the rate of photosynthesis? | It will increase at first, but if it gets too hot it will decrease and stop | | | | | | | | | | | | | | | | | | | |
| How can a leaf be tested for carrying out photosynthesis? | <ul style="list-style-type: none"> • Boil it in water to kill it • Put it into boiling ethanol to remove the chlorophyll (green colour) • Add iodine which will turn blue/black if starch is present | | | | | | | | | | | | | | | | | | | |
| What piece of equipment could be used for measuring the volume of gas produced during photosynthesis? | A gas syringe Or An unturned measuring cylinder filled with water | | | | | | | | | | | | | | | | | | | |
| Suggest three reasons that plants are so important to life on Earth. | <ul style="list-style-type: none"> • They produce oxygen which is essential for life on Earth • They provide biomass which is used by animals as food • They remove carbon dioxide from the atmosphere which prevents global warming and the Earth becoming too hot | | | | | | | | | | | | | | | | | | | |

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| <p>Name the labelled parts of the reproductive system in flower plants:</p> | A – stigma B – style C – ovary D – ovule E – anther F – filament |
| What is the male reproductive organ called in a plant? | Stamen |
| Which parts make up the male reproductive organ in a plant? | Anther and filament |
| What is the female reproductive organ called in a plant? | Carpel |
| Which parts make up the female reproductive organ in a plant? | Stigma, style, ovary and ovule |
| What is the name for the transfer of pollen to the stigma of a flowering plant? | Pollination |
| By which two main methods does pollination occur? | <ul style="list-style-type: none"> Insect pollination Wind pollination |
| What is the role of the petals in flowering plants? | To attract insects |
| What is the role of the sepals in flowering plants? | To protect the plant's reproductive system |
| What is the male gamete in plants? | Pollen |
| What is the female gamete in plants? | Eggs |
| Describe how fertilisation occurs in flowering plants. | Pollen travels from the stigma down the style. It then enters the ovule and combines with the egg. |
| What is formed following fertilisation of an egg cell? | A seed |
| What is the scientific word for ‘spreading out seeds’? | Dispersal |
| By which methods can seed dispersal take place? | <ul style="list-style-type: none"> By wind By animals By explosion By water |
| Why is it important for seeds to be dispersed? | To avoid competition for water/light/other resources |
| How are seeds which use dispersal by wind adapted? | They have a parachute or wings to allow them to travel further |

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| How are seeds which use dispersal by animals adapted? | <ul style="list-style-type: none"> • They have sweet flesh to encourage animals to eat them • A hard seed coat to avoid the seed being digested • Brightly coloured skin to attract animals | | | | | | | | | | |
| How are seeds which use dispersal by water adapted? | The outside (husk) is made of fibres which trap air. This helps them to float. | | | | | | | | | | |
| What three things are required for germination to occur? | <ul style="list-style-type: none"> • Water • Oxygen • Warmth | | | | | | | | | | |
| <p>Name the labelled parts of the germinating seed:</p>  | <p>A – food store B – seed coat C – shoot embryo D – root embryo</p> | | | | | | | | | | |
| What are the stages involved in germination? | <ul style="list-style-type: none"> • Water softens the seed coat • The food store dissolves in the water and reacts with oxygen, releasing energy • Roots and shoots start to form • Shoots break through the soil and can start to photosynthesise | | | | | | | | | | |

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| <p>What are the names of each labelled part of the male reproductive system:</p>  | <p>A – bladder B – penis C – sperm duct D – urethra E – testis F – scrotum G – foreskin</p> | |
| <p>What is the role of each of the following:</p> <ul style="list-style-type: none"> • Bladder • Sperm duct • Urethra • Testis • Scrotum | <ul style="list-style-type: none"> • Bladder – stores urine • Sperm duct – transports sperm from the testes to the urethra • Testis – produces and stores sperm • Scrotum – expands and contracts to control to temperature of the testis | |
| <p>What are the names of each labelled part of the female reproductive system:</p>  | <p>A – ovary B – oviduct (fallopian tube) C – uterus D – cervix E – vagina</p> | |
| <p>What is the role of each of the following:</p> <ul style="list-style-type: none"> • Ovary • Oviduct • Uterus • Cervix | <ul style="list-style-type: none"> • Ovary – develops and releases eggs • Oviduct – contains cilia (small hairs) which sweep eggs towards the uterus • Uterus – where the baby will develop • Cervix – holds the baby in place during pregnancy | |
| <p>What is the scientific term for ‘sex cells’?</p> | <p>Gametes</p> | |
| <p>In humans, what is the male gamete?</p> | <p>Sperm</p> | |
| <p>In humans, what is the female gamete?</p> | <p>Ovum (egg)</p> | |
| <p>What is the term used to describe the process of combining an ovum with a sperm cell?</p> | <p>Fertilisation</p> | |
| <p>What is the scientific term for a fertilised egg cell?</p> | <p>Zygote</p> | |

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| What is the potential impact of the mother drinking alcohol during pregnancy? | Premature birth, low birth weight and brain disorders | | | | | | | | | | |
| What is the potential impact of the mother smoking during pregnancy? | Premature birth, low birth weight and heart/breathing problems | | | | | | | | | | |
| How are waste products (e.g. carbon dioxide) excreted by the fetus? | The waste products travel through the umbilical cord, pass across the placenta, and are then excrete by the mother. | | | | | | | | | | |
| Whose blood flows inside the umbilical cord? | The fetus' | | | | | | | | | | |
| What changes take place in the body during puberty? | <ul style="list-style-type: none"> • Grow more body hair • Penis enlarges (in men) • Voice deepens (in men) • Menstrual cycle starts (in women) • Breasts develop (in women) • Hormones (testosterone in men and oestrogen in women are produced) | | | | | | | | | | |

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| What is a vaccination? | A weak or inactive form of the pathogen which triggers the production of lymphocytes in the body. This means that when the 'full' version of the pathogen enters the body, memory lymphocytes are already in the bloodstream. | | | | | | | | | |
| Suggest two things that we can personally do to act as a defence against disease. | <ul style="list-style-type: none"> • Maintain good hygiene (hand-washing, tooth brushing etc.) • Eat a balanced diet • Take regular exercise • Resting • Not smoking or drinking excessive volumes of alcohol | | | | | | | | | |
| What are the responsibilities of a community in preventing disease? | <ul style="list-style-type: none"> • Providing medical care • Removing rubbish • Providing safe drinking water • Maintaining high standard of health and hygiene in businesses | | | | | | | | | |

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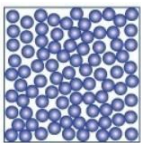
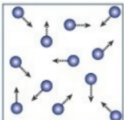
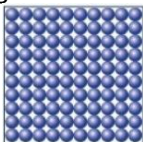
Biology – relationships in an ecosystem

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| What are some of the problems of deforestation? | <ul style="list-style-type: none"> • Habitat loss and extinction of species • Reduced soil fertility • Flooding and landslides • Changes to the atmosphere (less oxygen, more carbon dioxide, drier air) | | | | | | | | | |
| What are some conservation activities which may be carried out? | <ul style="list-style-type: none"> • Creation of new habitats – plants new trees, digging a garden pond • Creation of nature reserves • Captive breeding – such as in zoos | | | | | | | | | |
| What does the word 'biodiversity' mean? | A range of living organisms | | | | | | | | | |
| Why is biodiversity important? | Without biodiversity, it is more likely that the death of one species will result in the death of many more species | | | | | | | | | |

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| What are the names of the 3 states of matter? | Solid, liquid, gas | | | | | | | | | | | | | | | | | | |
| For which state of matter is this the particle diagram?  | Liquid | | | | | | | | | | | | | | | | | | |
| For which state of matter is this the particle diagram?  | Gas | | | | | | | | | | | | | | | | | | |
| For which state of matter is this the particle diagram?  | Solid | | | | | | | | | | | | | | | | | | |
| How are the particles arranged in a solid? | <ul style="list-style-type: none"> • Regular arrangement • Particles touching | | | | | | | | | | | | | | | | | | |
| How do particles move in a solid? | Vibrate about a fixed point | | | | | | | | | | | | | | | | | | |
| How are the particles arranged in a liquid? | <ul style="list-style-type: none"> • Random arrangement • Particles touching | | | | | | | | | | | | | | | | | | |
| How do particles move in a liquid? | Move around each other | | | | | | | | | | | | | | | | | | |
| How are the particles arranged in a gas? | <ul style="list-style-type: none"> • Random arrangement • Particles far apart | | | | | | | | | | | | | | | | | | |
| How do particles move in a gas? | Move freely | | | | | | | | | | | | | | | | | | |
| Explain why gases can be compressed, but solids and liquids cannot. | There is space between the particles, so they can be moved closer together. | | | | | | | | | | | | | | | | | | |
| Explain why gases and liquids can flow, but solids cannot. | The intermolecular forces in liquids and gases are weaker than in solids. This means that particles are not fixed in place. | | | | | | | | | | | | | | | | | | |
| What are intermolecular forces? | Forces between molecules | | | | | | | | | | | | | | | | | | |
| In which state of matter do the particles have most energy? | Gas | | | | | | | | | | | | | | | | | | |
| What causes gas pressure? | Collision of particles with the container wall | | | | | | | | | | | | | | | | | | |
| What is the term used for the random motion of particles? | Brownian motion | | | | | | | | | | | | | | | | | | |
| What is the definition for diffusion? | The movement of particles from an area of higher concentration to an area of lower concentration. | | | | | | | | | | | | | | | | | | |
| What type of change is a change of state? | Physical change | | | | | | | | | | | | | | | | | | |

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| What is the main difference between a chemical change and a physical change? | A chemical change results in new substances being formed, whereas a physical change does not | | | | | | | | | | | | | | | | | | |
| What are all the changes of state called? | Melting, freezing, evaporating, boiling, condensing and sublimating | | | | | | | | | | | | | | | | | | |
| What happens to the arrangement, movement and energy of particles during melting? | The particles gain energy, which means they move faster. This allows them to overcome the attractions between themselves enough to be able to move away from each other and out of their fixed positions. | | | | | | | | | | | | | | | | | | |
| What happens to the arrangement, movement and energy of particles during boiling/evaporation? | The particles gain energy, which means they move faster. This allows them to overcome the attractions between themselves enough to be able to move away from each other, which means they are no longer touching. | | | | | | | | | | | | | | | | | | |
| What happens to the temperature of a substance during evaporation? | It decreases | | | | | | | | | | | | | | | | | | |
| Explain why the average temperature of a substance decreases during evaporation. | The average energy of the particles in the substance has fallen (because the high energy particles have evaporated). | | | | | | | | | | | | | | | | | | |
| What state will a substance be if the temperature is above its boiling point? | Gas | | | | | | | | | | | | | | | | | | |
| What state will a substance be if the temperature is between its melting point and boiling point? | Liquid | | | | | | | | | | | | | | | | | | |
| What state will a substance be if the temperature is below its melting point? | Solid | | | | | | | | | | | | | | | | | | |
| What is the melting point of water? | 0°C | | | | | | | | | | | | | | | | | | |
| What is the boiling point of water? | 100°C | | | | | | | | | | | | | | | | | | |
| What happens to water when it freezes? | It expands | | | | | | | | | | | | | | | | | | |
| Why does water expand when it freezes? | The particles are further apart from each other | | | | | | | | | | | | | | | | | | |
| What does this mean happens to the density of water when it freezes? | It decreases (all other solids are denser than their liquid state) | | | | | | | | | | | | | | | | | | |
| What are the stages involved in the water cycle? | <ul style="list-style-type: none"> • Evaporation (from oceans and rivers) • Condensation (to form clouds) • Precipitation (as rain, snow etc.) • Run-off (water flows back to oceans and seas) | | | | | | | | | | | | | | | | | | |
| What can be done to increase the rate of evaporation? | <ul style="list-style-type: none"> • Better air flow (more wind) • Warmer temperatures • Larger surface area (shallower container) | | | | | | | | | | | | | | | | | | |

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| How could the volume of water lost over a number of days be accurately measured? | 1. Measure the mass of water before the experiment. 2. Measure the mass of water after the experiment. | | | | | | | | | | | | | | | | | | | |
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| What is the definition of the word 'atom'? | The smallest particle of a chemical element which can exist. | | | | | | | | | | | | | | | | | | | |
| What is the definition of the word 'molecule'? | Two or more atoms chemically joined together | | | | | | | | | | | | | | | | | | | |
| What is definition of the word 'compound'? | Two or more atoms of different types chemically joined together | | | | | | | | | | | | | | | | | | | |
| What is the definition of the word 'element'? | Atoms of the same type | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for hydrogen? | H | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for oxygen? | O | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for carbon? | C | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for nitrogen? | N | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for sulfur? | S | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for magnesium? | Mg | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for sodium? | Na | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for chlorine? | Cl | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for calcium? | Ca | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for copper? | Cu | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for iron? | Fe | | | | | | | | | | | | | | | | | | | |
| What is the chemical symbol for helium? | He | | | | | | | | | | | | | | | | | | | |
| What is the formula of a molecule of water? | H ₂ O | | | | | | | | | | | | | | | | | | | |
| What is the formula of a molecule of carbon dioxide? | CO ₂ | | | | | | | | | | | | | | | | | | | |
| What is the formula of a molecule of oxygen? | O ₂ | | | | | | | | | | | | | | | | | | | |
| What is the formula of a molecule of methane? | CH ₄ | | | | | | | | | | | | | | | | | | | |
| What is the formula of sodium chloride? | NaCl | | | | | | | | | | | | | | | | | | | |
| What is the formula of hydrochloric acid? | HCl | | | | | | | | | | | | | | | | | | | |
| What is the formula of sodium hydroxide? | NaOH | | | | | | | | | | | | | | | | | | | |
| What is the formula of calcium carbonate? | CaCO ₃ | | | | | | | | | | | | | | | | | | | |
| What is the formula of copper sulfate? | CuSO ₄ | | | | | | | | | | | | | | | | | | | |
| What is the formula of sulfuric acid? | H ₂ SO ₄ | | | | | | | | | | | | | | | | | | | |

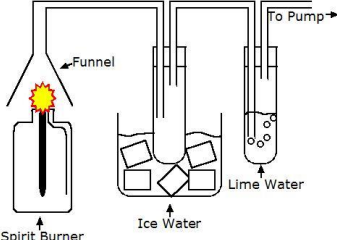
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|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|--|
| What is the definition of a pure substance? | A substance containing particles of only one type | | | | | | | | | | | | | | | | | | | |
| What is the definition of a mixture? | A substance containing particles of more than one type | | | | | | | | | | | | | | | | | | | |
| How can a pure substance be identified? | A pure substance melts and boils at a particularly temperature. A mixture melts and boils across a range of temperatures. | | | | | | | | | | | | | | | | | | | |
| What happens to the volume of most solids, liquids and gases when they are heated (with the exception of water)? | They expand | | | | | | | | | | | | | | | | | | | |
| How does a thermometer work? | The mercury or alcohol inside expands when it gets hot. This forces it up the capillary tube where the temperature can be read-off. | | | | | | | | | | | | | | | | | | | |
| What is the difference between evaporation and boiling? | Evaporation can happen at any temperature. Boiling occurs at a specific temperature for a particular substance. | | | | | | | | | | | | | | | | | | | |
| What is the law of conservation of mass? | Mass cannot be gained or lost because atoms cannot be made or destroyed | | | | | | | | | | | | | | | | | | | |
| What is a solvent? | A liquid into which a substance can be dissolved | | | | | | | | | | | | | | | | | | | |
| What is a solute? | A solid or a gas which has been dissolved | | | | | | | | | | | | | | | | | | | |
| What is a solution? | A mixture of a solvent and a solute | | | | | | | | | | | | | | | | | | | |
| What are three ways to increase the rate at which a substance will dissolve? | <ul style="list-style-type: none"> • Increase the temperature • Stir the solvent • Increase the surface area of the solute (grind it up!) | | | | | | | | | | | | | | | | | | | |
| What is the term used to describe a solution with only a small amount of solute dissolved? | Dilute | | | | | | | | | | | | | | | | | | | |
| What is the term used to describe a solution with a large amount of solute dissolved? | Concentrated | | | | | | | | | | | | | | | | | | | |
| What do we call a solution into which no more solute can be dissolved? | Saturated | | | | | | | | | | | | | | | | | | | |
| What is the effect of increasing the temperature upon the mass of solute which can dissolve in a solvent? | It increases | | | | | | | | | | | | | | | | | | | |
| What do we call a substance which cannot be dissolved in a solvent? | Insoluble | | | | | | | | | | | | | | | | | | | |
| What do we call a mixture of a solvent and an insoluble substance? | A suspension | | | | | | | | | | | | | | | | | | | |
| What are the two methods of separating an insoluble solid from a liquid? | <ul style="list-style-type: none"> • Decanting • Filtration | | | | | | | | | | | | | | | | | | | |
| What is decanting? | Allowing solid particles to sink to the bottom of a container (sedimentation) and then carefully pouring off the liquid | | | | | | | | | | | | | | | | | | | |

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| What is filtration? | Passing a suspension through a very fine sieve (normally made of paper). | | | | | | | | | | | | | | | | | | |
| How does filtration work? | Small, liquid particles, are able to pass through the pores in the filter paper. Larger, solid particles, get trapped and cannot pass through. | | | | | | | | | | | | | | | | | | |
| What is the name for the solid that is trapped by the filter paper? | Residue | | | | | | | | | | | | | | | | | | |
| What is the name for the liquid which passes through the filter paper? | Filtrate | | | | | | | | | | | | | | | | | | |
| What is the term used for the evaporation of a solvent to form crystals? | Crystallisation | | | | | | | | | | | | | | | | | | |
| What type of mixtures can be separated using simple distillation? | Mixtures of substances with different boiling points. Evaporation and condensation only happen once. | | | | | | | | | | | | | | | | | | |
| What type of mixtures can be separated using fractional distillation? | Mixtures of a number of substances with different boiling points. Evaporation and condensation happen several times. | | | | | | | | | | | | | | | | | | |
| What type of mixtures can be separated using paper chromatography? | A mixture of different coloured compounds dissolved in a liquid. These substances must have different levels of solubility. | | | | | | | | | | | | | | | | | | |
| How is paper chromatography carried out? | <ol style="list-style-type: none"> 1. A line is drawn in pencil towards the bottom of the chromatography paper 2. A small spot of the mixture is placed on the line 3. The bottom of the chromatography paper is placed in a solvent (usually water) and the water allowed to move up the paper | | | | | | | | | | | | | | | | | | |
| Why is the line drawn in pencil? | Graphite doesn't dissolve in water and so won't move up the paper | | | | | | | | | | | | | | | | | | |
| How high does the water level need to be? | Between the bottom of the paper and the pencil line | | | | | | | | | | | | | | | | | | |
| What is the equation for calculating the R_F value? | $R_F = \frac{\text{distance moved by solute}}{\text{distance moved by solvent}}$ | | | | | | | | | | | | | | | | | | |
| What does it mean if a spot doesn't move from the pencil line? | The substance doesn't dissolve in that solvent | | | | | | | | | | | | | | | | | | |
| What does the distance moved by a spot tell you about the solubility of the substance? | The further a spot moves, the more soluble it is | | | | | | | | | | | | | | | | | | |
| How can you tell the difference between pure and impure substances on a paper chromatogram? | A pure substance will only have one spot. An impure substance will separate into multiple spots | | | | | | | | | | | | | | | | | | |
| How can you tell if two substances from different mixtures are the same? | They will have the same R_F value (and will have travelled the same distance) | | | | | | | | | | | | | | | | | | |

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| <p>Describe how this equipment can be used to determine the products of combustion.</p>  | <p>Gases are collected by the funnel and passed through the gas-exchange tube. The ice water condenses the water vapour. The lime water turns cloudy due to the carbon dioxide.</p> | |
| <p>What is formed during the incomplete combustion of a hydrocarbon?</p> | <p>Carbon monoxide and soot (solid carbon particles)</p> | |
| <p>What is the problem with carbon monoxide?</p> | <p>It binds to your red blood cells preventing them from carrying oxygen around the body. This can lead to death.</p> | |
| <p>What is the problem with soot?</p> | <p>It makes buildings dirty and can cause problems for people with asthma (by irritating the trachea)</p> | |
| <p>Which human activities release carbon dioxide?</p> | <p>Any involving burning fossil fuels (driving petrol/diesel cars, making electricity etc.)</p> | |
| <p>What is the impact of carbon dioxide on the climate?</p> | <p>Carbon dioxide is a greenhouse gas and causes heat to be trapped inside the Earth's atmosphere</p> | |
| <p>How does the greenhouse effect work?</p> | <ul style="list-style-type: none"> • Light from the sun enters the atmosphere and hits the Earth. • The Earth absorbs and reemits some of this energy back into space. • Greenhouse gases absorb infrared radiation (heat) and reemit it back to Earth | |
| <p>How is sulfur dioxide produced?</p> | <p>Sulfur impurities in coal react with oxygen creating sulfur dioxide</p> | |
| <p>What is the problem with sulfur dioxide in the atmosphere?</p> | <p>Sulfur dioxide dissolves in clouds to create acid rain</p> | |
| <p>What is the problem with acid rain?</p> | <ul style="list-style-type: none"> • It corrodes buildings/statues • It kills fish and other aquatic organisms | |
| <p>How can the production of sulfur dioxide and carbon dioxide be reduced?</p> | <ul style="list-style-type: none"> • Burn fewer fossil fuels • Produce electricity using renewable methods • Drive electric cars (or walk/cycle) | |
| <p>What is a thermal decomposition reaction?</p> | <p>The breaking down of a substance using heat</p> | |
| <p>What are the products of the thermal decomposition of hydrated copper sulfate?</p> | <p>Dehydrated copper sulfate and water</p> | |

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| Which scale is used to measure the strength of acids and alkalis? | pH scale | | | | | | | | | | | | | | | | | | | |
| With universal indicator, what colour will a strong acid turn? What pH does this represent? | Red; 1-2 | | | | | | | | | | | | | | | | | | | |
| With universal indicator, what colour will a weak acid turn? What pH does this represent? | Yellow; 5-6 | | | | | | | | | | | | | | | | | | | |
| With universal indicator, what colour will a neutral substance turn? What pH does this represent? | Green; 7 | | | | | | | | | | | | | | | | | | | |
| With universal indicator, what colour will a weak alkali turn? What pH does this represent? | Blue/green; 8-9 | | | | | | | | | | | | | | | | | | | |
| With universal indicator, what colour will a strong alkali turn? What pH does this represent? | Purple; 13-14 | | | | | | | | | | | | | | | | | | | |
| What colour will litmus paper turn with an acid? | Red | | | | | | | | | | | | | | | | | | | |
| What colour will litmus paper turn with an alkali? | Blue | | | | | | | | | | | | | | | | | | | |
| How could you prepare an indicator using red cabbage, raw beetroot or blackcurrants? | <ul style="list-style-type: none"> Grind up the plant in water Filter the liquid Add to acid/alkali | | | | | | | | | | | | | | | | | | | |
| What is a better method for measuring pH, rather than using an indicator? | Using a pH probe | | | | | | | | | | | | | | | | | | | |
| What is the general word equation for the reaction between an acid and a base? | Acid + base → salt + water | | | | | | | | | | | | | | | | | | | |
| What is the general word equation for the reaction between an acid and a metal? | Acid + metal → salt + hydrogen | | | | | | | | | | | | | | | | | | | |
| What is the general word equation for the reaction between an acid and a metal oxide? | Acid + metal oxide → salt and water | | | | | | | | | | | | | | | | | | | |
| What is the general word equation for the reaction between an acid and a metal hydroxide? | Acid + metal hydroxide → salt + water | | | | | | | | | | | | | | | | | | | |
| What is the general word equation for the reaction between an acid and a metal carbonate? | Acid + metal carbonate → salt + water + carbon dioxide | | | | | | | | | | | | | | | | | | | |
| What is the name for the type of reaction between an acid and a base which forms a salt and water | Neutralisation reaction | | | | | | | | | | | | | | | | | | | |
| What is the method for making a pure salt from an acid and an insoluble base? | <ol style="list-style-type: none"> React an acid with excess base Filter the excess base Evaporate the water | | | | | | | | | | | | | | | | | | | |
| What is the effect of evaporating the water more slowly? | Larger crystals | | | | | | | | | | | | | | | | | | | |

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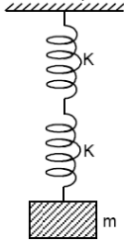
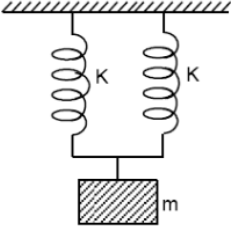
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| On a distance-time graph, what is represented by a flat line? | A stationary object | | | | | | | | | | | | | | | | | | |
| How can the speed of an object be calculated using a distance-time graph? | By calculating the gradient (steepness of the lines) – $\frac{\text{change in distance}}{\text{change in time}}$ | | | | | | | | | | | | | | | | | | |
| On a distance-time graph, what does a steep line represent? | Moving quickly | | | | | | | | | | | | | | | | | | |
| On a distance-time graph, what does a shallow line represent? | Moving slowly | | | | | | | | | | | | | | | | | | |
| In which direction does gravity act? | Towards the centre of mass (e.g. the centre of the Earth) | | | | | | | | | | | | | | | | | | |
| Which two factors do the strength of gravity depend upon? | <ul style="list-style-type: none"> The mass of both objects The distance between the objects | | | | | | | | | | | | | | | | | | |
| If the mass of the object increases, what happens to the size of gravity? | It increases | | | | | | | | | | | | | | | | | | |
| If the distance between the objects increase, what happens to the size of gravity? | It decreases | | | | | | | | | | | | | | | | | | |
| What is the meaning of the word 'mass'? | The amount of matter (stuff) that an object is made up of | | | | | | | | | | | | | | | | | | |
| What is the meaning of the word 'weight'? | A force caused by gravity acting upon a mass | | | | | | | | | | | | | | | | | | |
| What is the equation which links weight, mass and gravitational field strength? | $Weight = mass \times gravitational\ field\ strength$ | | | | | | | | | | | | | | | | | | |
| What are the units for mass? | Kilograms (kg) | | | | | | | | | | | | | | | | | | |
| What are the units for weight? | Newtons (N) | | | | | | | | | | | | | | | | | | |
| What are the units for gravitational field strength? | Newtons per kilogram (N/kg) | | | | | | | | | | | | | | | | | | |

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| If forces are balanced, what is the size of the resultant force? | Zero | | | | | | | | | | | | | | | | | | |
| If no resultant force acts upon an object, what will happen to its motion? | It will remain at a constant speed, in a constant direction (or will be stationary) | | | | | | | | | | | | | | | | | | |
| If two forces are acting in the same direction, how can the resultant force be calculated? | Add the forces together | | | | | | | | | | | | | | | | | | |
| If two forces are acting in opposite directions, how can the resultant force be calculated? | Take the smaller force away from the larger force | | | | | | | | | | | | | | | | | | |
| What is Hooke's law? | The amount of stretch for a spring is directly proportional to the mass added. | | | | | | | | | | | | | | | | | | |
| Which equation links: force, extension and spring constant | $Force = spring\ constant \times extension$ | | | | | | | | | | | | | | | | | | |

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| If the force applied to a spring is doubled, what will happen to the extension of the spring? | It will double | | | | | | | | | | | | | | | | | | |
| What is the term used for when a spring will no longer return to its original form? | Limit of proportionality | | | | | | | | | | | | | | | | | | |
| What is the term used for this arrangement of springs?  | Series | | | | | | | | | | | | | | | | | | |
| What is the effect on the total extension of the springs, of adding an identical spring in series? | The extension will double | | | | | | | | | | | | | | | | | | |
| What is the term used for this arrangement of springs?  | Parallel | | | | | | | | | | | | | | | | | | |
| What is the effect on the total extension of the springs, of adding an identical spring in parallel? | The extension will half | | | | | | | | | | | | | | | | | | |
| Which force opposes the forward motion of an object? | Friction (including air resistance and water resistance) | | | | | | | | | | | | | | | | | | |
| What causes air and water resistance? | The collision of gas (or water) molecules hitting an object. This exerts a force, slowing the object down | | | | | | | | | | | | | | | | | | |
| Which energy transfer happens as a result of friction? | Kinetic → thermal | | | | | | | | | | | | | | | | | | |
| What is meant by the term 'stopping distance'? | The distance required to stop a vehicle at different speeds | | | | | | | | | | | | | | | | | | |
| What is meant by the term 'thinking distance'? | The distance travelled between seeing a danger and applying the brake | | | | | | | | | | | | | | | | | | |
| What is meant by the term 'braking distance'? | The distance travelled between applying the brake and stopping | | | | | | | | | | | | | | | | | | |
| How is stopping distance calculated? | Stopping distance = thinking distance + braking distance | | | | | | | | | | | | | | | | | | |
| Which factors may affect the thinking distance? | <ul style="list-style-type: none"> • Speed of the vehicle • Visibility • Whether the driver has taken any drugs (alcohol, caffeine etc.) • Tiredness | | | | | | | | | | | | | | | | | | |

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| <p>Why does a drawing pin (see picture) go into the wall, but not hurt your thumb?</p>  | <p>The area of the pointed bit is small, and therefore the pressure is high. The area of the flat bit is large, and therefore the pressure small.</p> | | | | | | | | | |
| <p>Suggest 4 more examples of ways in which pressure is used in everyday life.</p> | <ul style="list-style-type: none"> • Studs on football boots sink into the ground • A sharp knife cuts things easily • A camel has a large foot to prevent it sinking into the sand • Large tractor tyres stop the tractor from sinking into the mud | | | | | | | | | |

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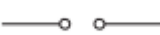







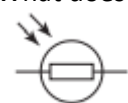


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| What could be the effects on the ear of hearing very loud sounds? | <ul style="list-style-type: none">• Perforated (broken) ear drum – temporary deafness• Damage to the cochlea – permanent deafness | | | | | | | | | |
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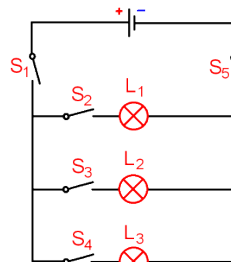
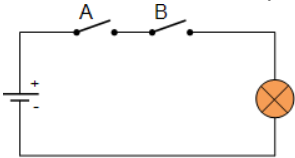
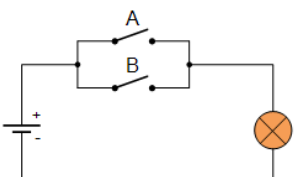
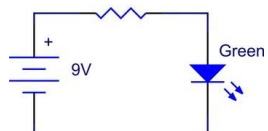
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| What is an electric current? | A flow of charged particles (electrons in wires) | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | A cell | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Terminals (ends of a wire) | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Buzzer | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Lamp/bulb | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Motor | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Open SPST switch | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Closed SPST switch | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Battery | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Fuse | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Light dependent resistor (LDR) | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Diode | | | | | | | | | | | | | | | | | | |
| What does this circuit symbol represent?  | Light emitting diode (LED) | | | | | | | | | | | | | | | | | | |

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| <p>In the parallel circuit below, what is the effect of opening switch 4 on each of lamp 1, 2 and 3 (assuming that all other switches are closed)?</p>  | <p>Lamp 1 will be on Lamp 2 will be on Lamp 3 will be off</p> | |
| <p>What is used to show the actions of switches in a circuit?</p> | <p>Truth tables</p> | |
| <p>What is the name used for this arrangement of switches? Why?</p>  | <p>An AND circuit because switch A and switch B must be closed for the lamp to light</p> | |
| <p>What is the name used for this arrangement of switches? Why?</p>  | <p>An OR circuit because switch A or switch B must be closed for the lamp to light</p> | |
| <p>What is a fuse used for?</p> | <p>Protecting electrical appliances from power surges.</p> | |
| <p>How does a fuse work?</p> | <p>If the current is too high, the wire inside the fuse will melt and break. This breaks the circuit.</p> | |
| <p>Which way should a diode (or LED) be placed in a circuit so that it works?</p> | <p>With the flat side of the triangle closest to the positive side of the cell (or battery)</p>  | |
| <p>Why must an LED be placed into a circuit the correct way around?</p> | <p>An LED has very low resistance in one direction and very high resistance in the other. This means that it will only work if placed the correct way around.</p> | |
| <p>What are the energy transfers which take place in a battery powered torch?</p> | <p>Chemical → electrical → light</p> | |
| <p>What is a short circuit?</p> | <p>When electrons take the easiest route to get back to the battery (e.g. if a piece of wire is placed in parallel with the bulb)</p> | |

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| Where is the strength of the magnetic field in a solenoid strongest? | In the centre of the coil | | | | | | | | | | |
| What is the effect of reversing the current? | The direction of the magnetic field will change | | | | | | | | | | |
| Suggest 4 uses of electromagnets. | <ul style="list-style-type: none"> • Electric bells • Picking up cars in a scrap-yard • Relay circuits • In magnetic door locks | | | | | | | | | | |

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Revisiting plan

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